# HIRSUTINOLIDES, GLAUCOLIDES AND SESQUITERPENE LACTONE FROM VERNONIA SPECIES

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Abstract—The investigation of nine Vernonia species afforded in addition to known sesquiterpenes 28 new ones. The structures were elucidated by high field <sup>1</sup>H NMR spectroscopy and the configurations were determined by NOE difference spectroscopy and, in one case, by X-ray analysis. The results indicated that configurations of several previously reported sesquiterpene lactones have had to be revised. In addition to known types two new ones, the jalcaguaianolides and the vernojalcanolides, are described. Furthermore some unusual reaction products are presented which, in part, led to some natural occurring lactones.

#### INTRODUCTION

From the large genus Vernonia (Compositae, tribe Vernonieae) with more than 1000 species many highly oxygenated sesquiterpene lactones have been reported [1]. Glaucolides and hirsutinolides are especially widespread. These sesquiterpene lactones with a 7,11-double bond and an oxygen function at C-13 so far only have been isolated from representatives of the tribe Vernonieae though some other types have been reported. We now have investigated two species from Peru, four species from Central America and three species from Brazil, as well as Pseudoelephantopus spicatus.

## RESULTS AND DISCUSSION

The aerial parts of Vernonia patens H.B.K., collected in Peru, afforded stilpnotomentolide 8-O-angelate (3), 8-desacylvernonataloide 8-O-angelate (4c), vernonataloide (4d) [6], 8 $\alpha$ -methacryloyloxycompactifloride (5a), the corresponding angeloyloxy derivative 5b, vernopatensolide 8-O-angelate (7a), 8 $\alpha$ -angeloyloxy-10 $\alpha$ -hydroxy-1-desoxyhirsutinolide 13-O-acetate (8b), 8 $\alpha$ -angeloyloxy-4 $\alpha$ ,5 $\alpha$ -epoxyjalcaguaianolide (10a), 8 $\alpha$ -angeloyloxy-4 $\alpha$ ,5 $\alpha$ -epoxy-13-O-methyljalcaguaianolide (10c) and 8 $\alpha$ -angeloyloxy-4 $\alpha$ ,5 $\alpha$ -epoxyjalcaguaianolide 13-O-acetate (10e).

The structure of 3 followed from the <sup>1</sup>H NMR spectrum (Table 1) which was close to that of the corresponding tiglate [2]. The relative position of the ester groups was established by mild reaction of 3 with potassium carbonate which led to 7a and 7b. The <sup>1</sup>H NMR spectrum of the latter (Table 1) clearly indicated a free 13-hydroxy group while the angelate at C-8 was still present. Most likely the formation of 7a, which also was isolated as a natural product, came about by way of a C-11 enolate, a

reaction which would have to be more rapid than saponification of the allylic acetate. The configuration at C-10 previously assigned to analogues of 3 (in lit. [2] compounds 1-6, 10 and 11, see Table 9) has to be changed as NOE difference spectroscopy showed that a 10αmethyl group was present. Thus clear NOEs were observed in the spectrum of 3 between H-10 and H-8 and between H-15, H-6 and H-8 (always in Table 8). Similarly in the spectrum of 7a NOEs were observed between H-14 and H-9\$, between H-5, H-10 and H-15, between H-10, H-5 and H-8 and between H-15, H-3 $\beta$  and H-5. The NOEs with H-5 further indicated that the  $\Delta^5$  double bond of lactone 7a had the Z-configuration. Although peaks in the spectrum of 7a at room temperature were very broad and even at 90° in benzene only a few of the signals became clear, all signals could be interpreted at  $-40^{\circ}$ . The data required a conformation with both H-5 and H-10 above and the keto group below the plane while the methyls were quasi-equatorial. This conformation did not allow hemiacetal formation which between the ketone and the C-4 hydroxyl group was observed in isomer 85 where the  $\Delta^5$ double bond had the E-configuration (see below).

The structure of 4c could be deduced easily from the  $^1H$  NMR spectrum (Table 2) which was close to that of the corresponding methacrylate [6, 7]. NOE difference spectroscopy led to a conformation with the methyls at C-4 and C-10 above the plane as in the crystalline state [7]. Accordingly, NOEs were observed between H-14, H-8 and H-9 $\beta$  and between H-15, H-6 and H-8 as well as between H-5 and H-3 $\alpha$ .

The <sup>1</sup>H NMR spectra of 5a and 5b (Table 2) showed that these lactones differed from 8 $\alpha$ -caproyloxycompactifloride [5] only by the ester groups. NOE difference spectroscopy showed that the  $\Delta^5$  double bond had the *E*-configuration. Clear effects were observed between H-14, H-8 and H-9 $\beta$  and between H-15 and H-8. Inspection of a

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model showed that this required a change in the configuration assigned to the  $\Delta^5$  double bond in some esters reported previously (compounds 4 and 5 in lit. [5]).

The lactones 10a, 10c and 10e differed only by the nature of the oxygen function at C-13. The presence of guaianolides was deduced from the results of spin decoupling which led to the sequences H-1-H-3 and H-8-H-9. As a singlet around  $\delta$ 5.30 (Table 3) obviously was due to H-6 all data agreed with the proposed structure which was further supported by NOE difference spectroscopy. Clear effects were observed between H-14, H-6 and H-8, between H-15 and H-6 and between H-6 and H-8. Acetylation of 10a and 10e gave the corresponding diacetates which excluded the possibility that the guaianolides were 4,5-dihydroxylated.

The aerial parts of *Vernonia jalcana* Cuatr. gave after repeated TLC and HPLC 2a, the vernonataloides 4a, 4b and 4d [6, 7], the glaucolides 6a [8] and 6b [8], the hirsutinolides 8c, 8d, 8e, 8f, 8g and 8h [9], the guaianolides 10b, 10d and 11, as well as four cadinanolides (12a-12d).

The structure of 2a, which most likely is the precursor of 4a, followed from the <sup>1</sup>H NMR spectrum in deuteriobenzene at 77° (Table 2) where spin decoupling was possible, while in deuteriochloroform at 20° most signals were broadened. The data were close to those of the corresponding 8-O-methacrylate [10]. Similarly the structures of 4a and 4b could be deduced easily from the <sup>1</sup>H NMR spectra (Table 2) which were close to that of 4c

(see above). The changed nature of the ester residue at C-8 clearly followed from the typical <sup>1</sup>H NMR signals.

The structure of **8b** followed from the <sup>1</sup>H NMR spectral data (Table 1) which was close to those of similar lactones which differ only in the nature of the ester groups [2, 4]. X-ray analysis of **8b** (see Fig. 1) showed that the 10-methyl group was  $\alpha$ -orientated and the ether ring was below the plane. Accordingly, this configuration has to be changed in esters reported previously [2, 4] (see Table 9).

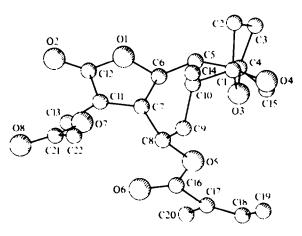


Fig. 1.

Table 1. <sup>1</sup>H NMR spectral data of sesquiterpene lactones 3, 7a, 7b, 8a and 8b (400 MHz, CDCl<sub>3</sub>, TMS as int. standard)

	3	7a ( -40°)	76	*	8a (C <sub>6</sub> D <sub>6</sub> , 77°)•
H-1					3.99 dd br
H-2	2.69 ddd	2.50 m	2.3-2.5 m	2.33 dd br	1.55-
H-2	2.64 ddd	2.36 m	23-23 M	_	} 1.55 m
H-3	2.30 ddd	2.36 m		_	_
H-3'	1.72 <b>ddd</b>	2.05 m	2.00 m	_	_
H-5	2.57 d	5.48 s	5.38 br	5.88 s	5.67 s
H-6	4.90 d br				
H-8	4.90 dd br	5.46 dd	5.50 d br	6.21 d br	6.37 d br
H-9	2.19 ddd	1.88 <i>đád</i>	1.90 m	_	1.83 d br
H-9'	2.53 ddd	2.48 dd br	2.40 m	_	2.35 dd
H-10	2.97 <b>dd</b> q	2.93 dq br	2.92 m	_	
H-13	4.90 d br	4.99 d	4.62 s br	5.09 ₫	5.22 d
H-13'	4.83 d br	4.95 d	4.02 S DF	5.02 d	5.08 d
H-14	1.14 <i>d</i>	1.10 d	1.08 d	0.94 d	0.98 s†
H-15	1.62 s	1.56 s	1.51 s <i>br</i>	1.47 s	1.14 s†
OAc	2.07 s	2.06 s		2.07 s	1.71 s
OR	6.22 <i>qq</i>	6.28 <i>qq</i>	6.21 <i>qq</i>	6.08 qq	5.76 qq
	1.99 dq	2.05 dq	2.03 dq	1.98 dq	1.92 dq
	1.87 dq	1.91 dq	1.96 dq	1.89 dq	1.83 dq
ОН	-	3.18 s	3.05 s	-	•

<sup>\*</sup>In CDCl<sub>3</sub>: H-2\beta 2.55 m; H-3\beta 2.41 m.

Table 2. <sup>1</sup>H NMR spectral data of sesquiterpene factores 2a, 4a-4c, 5a and 5b (400 MHz, CDCl<sub>3</sub>, TMS as int. standard)

	2a (C <sub>6</sub> D <sub>6</sub> , 77°)	4a	4	<b>4</b> c	5a	56
H-1	4.90 m	2.68 d br	2.68 d br	2.69 d br	3.06 dd	3.06 dd
H-2	•	2.11 dd br	2.13 dd br	2.13 dd br	2.15 m	2.15 m
H-2	•	1.52 m	1.55 m	1.55 m	1.57 m	1.56 m
H-3	•	2.29 ddd	2.30 <i>ddd</i>	2.31 <i>ddd</i>	2.28 ddd	2.28 ddd
H-3°	•	1.33 <i>ddd</i>	1.34 ddd	1.35 <b>ddd</b>	2.07 <b>ddd</b>	2.10 ddd
H-5	2.07 dbr	2.57 d	2.584	2.59 d	6.38 s	6.39 s
H-6	4.62 d br	4.92 d br	4.94 d br	5.02 d br	_	_
H-8	4.98 dd	5.17 d br	5.20 d br	5.25 d br	5.91 d	5.91 d
H-9	2 60	2.68 d br	2.66 d br	2.75 d br	2.62 d	2.60 d
H-9	2.50 m	1.97 dd br	2.01 dd br	2.03 dd br	1.88 dd	1.87 dd
H-13	6.18 dd	5.02 dd	5.01 dd	4.91 dd	5.02 d	5.00 d
H-13'	4.95 dd	4.77 dd	4.79 dd	4.80 dd	4.89 d	4.90 d
H-14	1.38 s br	1.50 s	1.52 s	1.54 s	1.48 s	1.49 s
H-15	1.04 s br	1.46 s	1.47 s	1.50 s	1.78 s	1.78 s
OAc	1.76 s	2.06 s	2.10 s	2.02 s	2.04 s	2.06 s
OR	1.65 s	2.10s	2.56 qq	6.24 <i>qq</i>	6.16 s <i>br</i>	6.23 qq
			1.18 d	2.00 dq	5.72 <b>dq</b>	2.02 dq
			1.14 d	1.87 dq	1.96 dd	1.90 dq

<sup>\*</sup>Unassigned multiplets.

<sup>†</sup> May be exchangeable.

<sup>-,</sup> Unassigned multiplets.

J (Hz): Compound 3: 2, 2' = 15; 2, 3 = 4.5; 2, 3' = 9.5; 2', 3 = 2', 3' = 6; 3, 3' = 13.5; 5, 6 = 9; 8, 9 = 3.5; 8, 9' = 5; 9, 9' = 15.5; 9, 10 = 3.5; 9', 10 = 11.5; 10, 14 = 7; 13, 13' = 13; compounds 7a and 7b: 8, 9 = 2.5; 8, 9' = 11.5; 9, 9' = 13.5; 9, 10 = 11; 10, 14 = 7; 13, 13' = 12.5; compound 8a: 1, 2 = 7; 1, 2' = 9; 8, 9' = 8; 9, 9' = 15; 13, 13' = 13; compound 8b: 2, 2' = 16.5; 2, 3 = 12.5; 8, 9' = 7; 10, 14 = 7; 13, 13' = 13.

J (Hz): Compound 2a: 5, 6 = 8.5; 6, 13 = 6, 13' = 1; 8, 9 = 4; 8, 9' = 10.5; 13, 13' = 12.5; compounds 4a, 4b and 4c: 1, 2' = 10.5; 2, 2' = 14; 2, 3 = 2; 2, 3' = 6; 2', 3 = 5; 2', 3' = 3, 3' = 13.5; 5, 6 = 8.5; 8, 9' = 9; 9, 9' = 14; 13, 13' = 13; compounds 5a and 5b: 1, 2 = 2.5; 1, 2' = 10.5; 2, 3' = 4.5; 2', 3' = 11; 3, 3' = 14.5; 8, 9' = 9.5; 9, 9' = 14; 13, 13' = 13.

Table 3. <sup>1</sup> H NMR spectral data of sesquiterpene lactones 10a-10e, 10e-acetate and 11 (400 MHz,
CDCl <sub>3</sub> , TMS as int. standard)

	10a	10Ь	10c	10 <b>d</b>	10e	10e-acetate	11
—— Н-1	2.32 d br	2.28 d br	2.30 d br	2.31 d br	2.31 d br	2.30 d br	2.62 m
H-6	5.28 s	5.24 s	5.30 s	5.26 s	5.32 s	5.34 s br	
H-8	5.76 dd	5.63 dd	5.70 dd	5.72 <b>dd</b>	5.76 dd	5.84 dd br	5.33 dd
H-9	1.49 dd	1.49 dd	1.50 dd	1.49 dd	1.49 dd	1.50 dd br	1.50 dd
H-9′	2.64 dd	2.59 dd	2.63 dd	2.57 dd	2.63 dd	2.61 <i>dd</i>	2.32 m
H-13 }	4.58 s	4.32 d	4.26 d	5.09 d	4.98 d	5.00 d br	4.35 d
H-13′ ∫	4.38 \$	4.20 d	4.22 d	4.88 d	4.87 d	4.86 d br	4.23 d
H-14	1.43 s	1.43 s	1.43 s	1.43 s	1.43 s	1.65 s	1.31 s
H-15	1.50 s	1.46 s	1.49 s	1.47 s	1.49 s	1.50 s	1.53 s
OAc		2.16 s		2.15 s	2.03 s	2.11 s	
				2.07 s		2.01 s	
OMe		3.34 s	3.30 s				3.40 s
OR	6.26 qq		6.24 <i>qq</i>		6.26 <i>qq</i>	6.21 qq	2.62 m
	2.04 dq		2.04 dq		2.04 dq	2.02 dq	1.21 d
	1.97 dq		1.97 dq		1.92 dq	1.89 da	1.24 d

J (Hz): Compounds 10a-10e: 1, 2 = 8; 8, 9 = 10; 8, 9' = 2; 9, 9' = 13; 13, 13' = 12; compound 11: 8, 9 = 3.5; 8, 9' = 12; 9, 9' = 12; 13, 13' = 11.

Most likely identical stereochemistry at C-4 is present in all similar lactones from *Vernonia* (see below). Furthermore the conformation indicated that the downfield shift of the H-8 signal in hirsutinolide is not caused by a deshielding effect of the ether oxygen but is due to the effect of the 7,11-double bond.

The structure of 8a also could be deduced from the <sup>1</sup>H NMR spectrum (Table 1) which was close to that of the corresponding capronate [5]. NOE difference spectroscopy indicated that the stereochemistry at C-10 differed from that of 8b as the methyl group in 8a was  $\beta$ -orientated. Accordingly, clear NOEs were observed between H-14, H-1, H-2 $\beta$ , H-8 and H-9 $\beta$ , and between H-8, H-2 $\beta$  and H-3 $\beta$ . The <sup>1</sup>H NMR spectra of the hirsutinolides 8c-8g (Table 4) again only could be interpreted at elevated temperature. Spin decoupling and the signals of the oxygen functions indicated that we were dealing with lactones which were closely related to 8b [9]. The presence of a free 13-hydroxyl group in 8c followed from the

upfield shift of the H-13 signal if compared with that of 8h. As 8d could not be acetylated under mild conditions the methoxy group was at C-13. The same was true in case of 8f with an additional methoxy group at C-1. Comparison of the <sup>1</sup>H NMR spectra of 8e with those of 8d and 8g with 8f clearly showed that these compounds were the corresponding methacrylates. As already discussed in the case of 8b (see above) all hirsutinolides most likely have the same stereochemistry which now is represented in an uniform manner following the conventions enunciated by Rogers et al. [11]. Most likely in all cases the configurations at C-1, C-4 and C-8 are identical. The methyl group at C-10 mostly is  $\alpha$ -orientated but changed to  $10\beta$  if an oxygen function is present at C-10 (Table 9).

The formulae of many previously reported hirsutinolides have to be revised (in lit. [4] compounds 21-23, 29-31 and 36, in lit. [5] compounds 7 and 8, in lit. [9] compounds 8-13, in lit. [12] compounds 13a-13e and 14, in lit. [14] compounds 3, 4 and 7, in lit. [15] compounds

Table 4. HNMR spectral data of sesquiterpene lactones 8c-8g

		8c		84	8f	8e	;	Be
	CDCl <sub>3</sub>	$C_{\bullet}D_{\bullet}(75^{\circ})$	CDCl <sub>3</sub>	C <sub>6</sub> D <sub>6</sub> (72°)	CDCI,	CDCI3	CDCl <sub>3</sub>	C <sub>6</sub> D <sub>6</sub> (72°)
H-5	5.92 s br	5.62 s	5.86 s br	5.66 s	5.91 s br	5.84 s br	5.88 s br	5.55 s
H-8	6.45 m	6.11 d br	6.36 m	6.31 d br	6.29 m	6.56 d br	6.50 m	6.62 dd
H-9	2.58 m	2.14 dd	2.53 m	2.36 dd	2.46 d br	2.58 d br	2.51 dd	2. <b>44</b> dd
H-9′	2.12 m	1.69 d br	2.07 m	1.92 dd	2.07 d br	2.09 d br	2.09 d br	1.96 dd
H-13		4.55 d	4.50 d br	4.48 d	4.47 d br	4.58 d br	4.53 d br	4.53 d
H-13' }	4.61 s br	4.45 d	4.24 d br	4.24 d	4.23 d br	4.27 d br	4.28 d br	4.31 d
H-14 Ć	1.60 s br	1.36 s	1.57 s	1.30 s	1.62 s	1,57 s br	1.61 s	1.29 s
H-15	1.26 s br	1.09 s	1.22 s br	1.12 s	1.24 s	1.22 s br	1.25 s	1.19 s
OAc	2.14s	1.60 s	2.10 s	1.70 s	2.11 s		-	_
OMe	_		3.38 s	3.17 s	3.51 s	3.40 s	3.55 s	3.49 s
		_		_	3.38 s		3.40 s	3.16 s
OR		-			-	6.28 s br	6.33 s br	6.29 s br
						5.67 <b>dq</b>	5. <del>69</del> <b>44</b>	5.31 dq
						1.94 s br	1.97 s br	1.86 s br

	8a	8b	8c	8d	8e	18	8g	8h	8i	8k	81
R	Ac	Ac	Н	Me	Me	Me	Me	Ac	Ac	Ac	Ac
R1	Ang	Ang	Ac	Ac	Meacr	Ac	Meacr	Ac	Ac	Ac	Meacr
$\mathbf{R}^{2}$	Me	н	Me	Me	Mc	Mc	Me	Me	Me	Me	Mc
R1	ОН	Mc	OH	OH	OH	OH	OH	OH	OAc	OAc	OH
R*	н	OH	он	ОН	ОН	OMc	OMe	ОН	OH	OMe	OH

3-7, in tit. [16] compounds 1 and 2, in lit. [17] compounds 1g and 1h, in lit. [18] compounds 1s-1f) (see Table 9).

The spectrum of 10b was very similar to that of 10c and that of 10d was close to that of 10e as in these pairs of lactones the 8-O-angelates were replaced by 8-O-acetates (Table 3). Another guaianolide 11 most likely had been formed via the corresponding 4,5-epoxide of type 10, which, however, was not isolated. The <sup>1</sup>H NMR spectrum (Table 3) fully agreed with the proposed structure. The configuration at C-4 was deduced from the chemical shift of H-15 and from biogenetic considerations, while that of the other chiral centres obviously was the same as in the lactones of types 10a-10e where the stereochemistry was established by NOE difference spectroscopy.

The structural elucidation of 12n-12d caused some problems. However, in deuteriobenzene all signals of the acetate of 12b could be assigned by spin decoupling (Table 5). The absence of a proton at C-6 and a lowfield singlet at δ6.28 as well as the couplings of H-2, H-3, H-8 and H-9 indicated that cadinanolides were present. Biogenetic considerations also supported this assumption as these lactones most likely are formed via enol lactones of type 12e. The position of the free hydroxyls was deduced from the shift differences in the <sup>1</sup>H NMR spectra of 12b and its acetylation product (Table 5). As the H-15 signals were shifted downfield a 4-0-acetate was formed while acetylation of the hydroxy at C-1 most likely was sterically hindered. The stereochemistry was determined by NOE difference spectroscopy. Clear NOEs were

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Table 5. 1H NMR spectral data of sesquiterpene lactones 12a-12d (400 MHz, CDCl<sub>3</sub>, TMS as int. standard)

	12a	12a- acetate	12 <b>b</b>	12b- acetate	$(C_4D_4)$	12c	12d- acetate
H-2	2.36 m	2.36 m	2.38 m	2.35 ddd	1.88 dd	2.39 m	2.38 m
H-2'	1.65 m	1.73 m	1.67 m	1.72 m	1.32 <b>ddd</b>	1.67 m	1.74 m
H-3	2.30 m	2.41 <i>ddd</i>	2.33 m	2.41 ddd	2.30 ddd br	2.34 m	2.38 m
H-3'	1.85 m	2.70 <b>ddd</b>	1.86 m	2.69 <b>ddd</b>	2.65 <b>ddd</b>	1.86 m	2.72 m
H-5	5.85 s	6.14 s	5.92 s	6.14 <i>s</i>	6.28 s	5.91 s	6.13 s
H-8	5.82 dd	5.85 dd	5.87 <b>dd</b>	5.85 dd	5.87 44	5.81 <i>dd</i>	5.70 <b>dd</b>
H-9	3.30 dd	3.28 dd	3.32 dd	3.27 dd	3.25 dd	3.36 dd	3.43 dd
H-9′	2.05 dd	2.06 dd	2.03 dd	2.06 dd	1.96 dd	2.05 dd	2.12 44
H-13	4.55 d	5.12 <i>d</i>	4.41 d	4.40 d	4.54 d	4.47 d	4.58 d
H-13'	4.47 d	4.79 d	4.23 d	4.21 d	4.26 d	4.23 d	4.27 d
H-14	1.71 s	1.70 s	1.70 s	1.70 s	1.58 s	1.70 s	1.71 s
H-15	1.40 s	1.77 s	1.40 s	1.77 s br	2.05 s br	1.41 s	1.77 s
OAc	2.21 s	2.21 s	2.21 s	2.20 s	1.98 s	2.21 s	2.15 s
	1.98 s	2.07 s	1.97 s	2.06 s	1.96 s	1.96 s	1.92 s
	2.06 s	2.04 s	2.03 s	1.93 s	1. <del>69</del> s		1.89 s
		1.94 s	_	1.91 s	1.67 s		_
	_	1.92 s	_		_		
OMe	_	_	3.34 s	3.33 s	3.15 s	3.34 s	3.34 s
OR	_	_		_	_	2.54 qq	5.99 dq
						1.17 d	5.62 da
						1.08 ₫	1.95 s bi

J (Hz): 2, 2' = 15; 2, 3 = 4.5; 2, 3' = 3; 2', 3 = 15; 2', 3' = 5; 3, 3' = 14; 8', 9 = 2; 8', 9' = 4; 13, 13' = 11.5; 9, 9' = 15.

observed with 12b between H-15 (assigned by a W-coupling with H-3) and H-2 $\beta$  and between H-3 $\alpha$  and H-5. The cis-annealation of the cyclohexane rings was deduced from the NOEs between H-14 and both H-9 protons and a NOE with H-2 $\alpha$ . The axial orientation of the ester group at C-8 followed from the small couplings  $J_{0,0}$ . The <sup>13</sup>C NMR spectra of 12b and its acetate (see Experimental) further supported the proposed structures. The structures of 12a, 12c and 12d, which also were transformed to their 4-O-acetates, were deduced from the <sup>1</sup>H NMR spectra which were close to that of 12b and its acetate respectively (Table 5). The changed nature of the oxygen functions followed from the typical <sup>1</sup>H NMR signals.

A reinvestigation of Vernonia cotoneaster Less. gave 1, vernonallenolide (9c) [19, 20] as well as two further allenes, the diol 9n and the monoacetate 9h. The structures of 9n and 9h followed from their <sup>1</sup>H NMR spectral data (Table 6) which were close to those of 9c [19, 20]. As expected the signals of H-8 and H-13 in 9a were shifted to higher field, and in the spectrum of 96 the position of the acetoxy group also could be deduced from the chemical shifts of these signals. The previously unknown configurations at C-2 and C-10 could be assigned on the basis of NOEs between H-14 and H-9a, between H-8 and H-15 and between H-5 and H-3 $\alpha$ , and the  $\Delta^4$  double bond has the E-configuration (NOE between H-6 and H-15). The <sup>1</sup>H NMR spectrum of 1 (Table 6) showed that this lactone was the 8,13-bis-desacetyl derivative of a triacetate isolated previously from the same species [5] whose C-3 stereochemistry originally assumed to be a, was subsequently corrected to  $\beta$  by NOE difference spectroscopy [6]. The position of the acetoxy group in the lactone 1 followed from the chemical shift of H-2.

Some additional Vernonia species gave only known lactones. Vernonia acuminata Less., V. arctioides Less. and

Table 6. <sup>1</sup>H NMR spectral data of sesquiterpene lactones 1, 9a, 9b and 13 (400 MHz, CDCl<sub>3</sub>, TMS as int. standard)

	1	9a	96	13
——— H-I	5.16 <i>d</i> br	_	_	
H-2	5.73 ddd br	5.32 dddg	5.38 <i>dddq</i>	2.18 m°, 3.36 m°
H-3	2.23 dd	2.28 dd	2.30 dd	2.34 m <sup>4</sup>
H-3'	2.78 dd	2.87 dd	2.89 dd	1. <b>80 m°</b>
H-5	4.49 dq	4.60 d br	4.66 d br	3.41 d
H-6	5.67 d br	5.71 <b>ddd</b>	5.89 d br	4.80 d
H-8	4.14 m	4.14 ddd	4.95 dd	5.23 dd
H-9	2.44 dd	2.24 dd	2.24 dd	2.18 dd
H-9'	2.94 dd br	2.9 <b>4 ddd</b>	2.90 ddd	3.17 <b>dd</b>
H-13 )		4.66 d br	4.55 d br	6.65 s
H-13' }	4.71 s br	4.72 d br	4.52 dd	5.99 s
H-14	1.55 s	1.66 d	1.69 d	1.46 s
H-15	1.81 d	1.95 d	2.01 d	1.35 s
OAc	2.11 s	_	2.08 s	
ОН	_	3.84 d	2741br	_
OCOR	_	_	_	6.07 <b>dq</b>
				5.62 dq
				1.88 đđ

<sup>\*</sup>Not first order.

J (Hz): 1. 1, 2 = 2, 3' = 8; 2, 3 = 2.5; 3, 3' = 13; 5, 6 = 10; 5, 15 = 1; 8, 9 = 10.5; 8, 9' = 5; 9, 9' = 12.5; compounds 9a and 9b: 2, 3 = 7; 2, 3' = 2.5; 2, 9' = 2, 14 = 3; 3, 3' = 13.5; 5, 6 = 9; 5, 15 = 1; 8, 9 = 11; 8, 9' = 5; 9, 9' = 13; 13, 13' = 13; (9a: 6, 13 = 6, 13' = 1; 8, OH = 8); compound 13: 5, 6 = 9.5; 8, 9 = 7.5; 8, 9' = 7.5; 9, 9' = 13.5.

V. brachiata Benth. ex Oerst. afforded 6a [8], while V. aurea Mart ex DC. contained 6a, 8i [9] and 8k [15]; a reinvestigation of V. mariana Mart. [12] gave 4d [6], 5a and 11β,13-dihydroxy-11,13-dihydroeremanthin [21] and V. greggii A. Gray 6b as previously [22], and 8l [18].

Pseudoelephantopus spicatus (B. Juss.) Rohr. gave 2a, 6a and 13. The structure of 13 followed from the <sup>1</sup>H NMR spectral data (Table 6) which were in part similar to that of 16b (see below). The absence of couplings from the signal of H-13 required substitution at C-7. Spin decoupling allowed the assignment of most signals though those of H-2 and H-3 were not first order multiplets. Obviously 13 was formed via the 10-O-desacetyl derivative of 6b by an addition-elimination process starting with an alcoholate ion of the 10-hydroxy group.

As mentioned above the 4,5-epoxyglaucolide 3 was converted easily to 7a by mild treatment with potassium carbonate. Similarly 4c was transformed probably via 5b to 14 and 10e. The structure of 14 was deduced from the <sup>1</sup>H NMR spectral data (Table 7). All signals could be assigned by spin decoupling and the stereochemistry followed from the couplings observed and from comparison of the spectra with those of closely related lactones. The lactones 14 and 10e most likely were formed

by reaction of the 5,6-double bond of 5b with the epoxide carbons C-1 and C-2 respectively. While the corresponding decalin ion then is transformed by loss of a proton to 14, the formation of 10e required a hydride migration from C-5 to C-6 followed by addition of the 4-hydroxy oxygen at C-5.

Compounds 6a and 8i reacted readily with base, both giving 16b, while 6a most likely was first transformed to an intermediate 15 by opening of the epoxide and Transesterification hemiacetal formation. addition elimination at the  $\Delta^{7(11)}$ -double bond then presumably led to 16a which was hydrolysed to 16b (see Scheme 1). The structure of 16b was deduced from the <sup>1</sup>H NMR spectrum (Table 7) and by NOE difference spectroscopy. Clear NOEs between H-8 $\beta$  and H-3 $\beta$ , between H-14 and H-9 and between H-15 and H-5 required the proposed configurations. The absence of a keto band in the IR spectrum and the molecular formula further supported the structure. The high reactivity of the 4,5-epoxides also could be observed in the case of 2b and 20 [20]. Base treatment of 20 gave the allene carbinol 21 [20] and the isomeric 1,2-dehydro bourbonenolides 22b and 23 as well as the corresponding desacetyl derivative 22a. The structures of these products followed from their

16n R = Ac 16b R = H

Scheme 1.

rable 1. The NMTR spectral data of sesquiterpen	: lactones 14, 166, 17 19, 228, 226 and 23 (400 MHz, CDCl <sub>3</sub> , 1 MS as int. standard)

			17a	$(C_{\bullet}D_{\bullet})^{\bullet}$		17 <b>b</b> †					
	14	166	Conf. I	Conf. II	Conf. I	Conf. II	18	19	22a	226	23
—— Н-1	3.40 dd		4.55 d br	4.53 d br	4.83 d br	4.86 d br	2.38 d br	2.2 <b>4</b> ddd			
1-2	2.04 m	} 2.11 m	2.6 <b>4</b> dddd	<b>‡</b>	2.76 dddd	2. <b>29 ddd</b> d	<b>‡</b>	<b>‡</b>	5.56 ddd	5.56 <b>ddd</b>	5.88 <b>ddd</b>
1-2'	1.58 m	} 2.11 m	<b>‡</b>	<b>‡</b>	<b>‡</b>	<b>‡</b>	<b>‡</b>	<b>‡</b>	_	_	_
1-3	1.00-	2.29 m	<b>‡</b>	<b>‡</b>	<b>‡</b>	<b>‡</b>	<b>‡</b>	<b>‡</b>	2.82 dd br	2.82 ddd br	2.75 d br
H-3'	} 1.90 m	2.01	<b>‡</b>	<b>‡</b>	<b>‡</b>	<b>‡</b>	<b>‡</b>	<b>±</b>	2.44 ddd	2.47 ddd	2.35 ddd
1-5	_	5.42 s	5.81 s	4.63 s	5.69 s	5.01 s	2.69 d	2.42 d	3.91 <b>ddd</b>	3.93 <b>ddd</b>	3.52 dd bi
H-8	6.11 <i>dd</i>	5.52 dd	6.34 <i>dd</i>	5.51 <b>dd</b>	6.34 dd	5.52 dd	6.35 dd br	6.14 dd br	4.81 ddd	5.95 ddd br	6.29 dd bi
1-9	2.28 dd	2.78 dd	2.59 dd br	2.16 dd	2.83 <b>dd</b>	2.36 dd	2.59 dd	2.43 dd	2.58 dd	2.52 <b>dd</b>	2.66 dd
1-9	2.21 dd	2.01 dd	2.04 d br	2.39 d br	2.23 d br	2.63 d br	1.94 dd	1.85 <b>dd</b>	2.39 dd	2.43 dd	2.08 dd
H-13	4.98 d	6.41 s	4.94 d	5.21 d	4.62 d br	1 400 1	4.89 d br	4.86 dd	5.15 d br	4.98 dd	4.93 dd
H-13°	4.78 d	5.85 s	4.91 d	5.00 d	4.56 d br	} 4.69 s br	4.75 d br	4.75 d br	4.94 d br	4.75 d br	4.81 dd
H-14	1.16 s	1.57 s	1.86 dd	1.81 <b>dd</b>	1.92 <b>dd</b>	1.81 <i>dd</i>	1.09 s	1.26 s	1.44 s	1. <b>44</b> s	1.51 s
H-15	1.49 s	1.55 s	1.42 s	1.16 s	1.71 s	1.48 s	1.22 s	1.38 s	1.13 s	1.16 s	1.61 s
OAc	1.98 s	2.00 s	1.68 s	1.65 s	_	_	2.05 s	2.01 s	2.07 s	2.15 s	2.07 s
							_	-	_	1.98 s	2.08 s
ЭН			<b>‡</b>	<b>‡</b>	<b>‡</b>	<b>‡</b>	2.95 s	<b>‡</b>	4.89 d	2.03 s	2.02 s
OR	6.13 <i>qq</i>	_	6.95 qq	•	6.92 qq	7.01 <i>qq</i>	6.91 qq	6.89 <i>qq</i>		=	
	1.94 qq		1.76 dg		1.88 <b>dq</b>	1.93 da	1.86 s br	1.84 da			
	1.83 dq		1.33 dq		1.87 dq	1.91 da	1.83 d br	1.82 d			

<sup>\*</sup>Ca 9:1.

<sup>†</sup>Ca 3:2.

**<sup>‡</sup>Unassigned multiplets.** 

J (Hz): Compound 14: 1, 2 = 7.5; 1, 2' = 11.5; 8, 9 = 9; 8, 9' = 4.5; 9, 9' = 14.5; 13, 13' = 13; compound 16b: 8, 9 = 7.5; 8, 9' = 4.5; 9, 9' = 14; compounds 17a and 17b: 1, 2 = 2, 2' = 2, 3' ≈ 13; 1, 14 = 1; 2, 3 = 2; 13, 13' = 13; Conf. I: 8, 9 = 5.5; 8, 9' = 2.5; 9, 9' = 14; Conf. II: 8, 9 = 2; 8, 9' = 11; 9, 9' = 12; compound 18: 1, 2 = 1, 5 = 7; 8, 9 = 9; 8, 9' = 7; 13, 13' = 12.5; compound 19: 1, 2 = 2, 1, 2' = 4; 1, 5 = 6; 8, 9 = 9; 8, 9' = 4; 8, 13 = 1; 9, 9' = 15.5; 13, 13' = 13; compounds 22a and 22b: 2, 3 = 2, 3' = 3, 5 = 3; 5 = 3; 2, 5 = 2; 3, 3' = 16.5; 8, 9 = 11; 8, 9' = 7.5; 8, 13 = 1.5; 9, 9' = 14.5; 13, 13' = 12.5; (compound 22b: 8, OH = 6.5); compound 23: 2, 3 = 1.5; 2, 3' = 3.5; 2, 5 = 3; 3, 3' = 16.5; 3, 5 = 2.5; 3', 5 = 0.5; 8, 9 = 8.5; 8, 9' = 8; 8, 13 = 1.5; 9, 9' = 14.5; 13, 13' = 13.

irr. of	3	<b>4</b> c	56	7 <b>a</b>	Sa	<b>9c</b>	10a	126	166	17a(l)	17a(II)	18	19	21	225	23
H-5		H-3 (7)		H-10 (6) H-15				H-3α (6)			H-8 (15)	H-1 (8) H-8 (12) H-15		H-3α (6) OH		
H-6						H-8 (5) H-15 (8)	H-8 (10)					*****				
H-8					H-2 <i>\$</i> H-3 <i>\$</i>	H-6 (5) H-15 (5)			H-3 <b>\$</b> (10)		H-5 (12)	H-1 (6) H-5 (12)		H-15	H-15	
H-10	H-8 (10)			H-5 (8) H-8 (6)												
H-14		H-8 (12) H-9 (5)	H-8 (6) H-9 (7)	(0)	H-1 (7) H-8 (8)	H-9α (6)	H-6 (8) H-8 (8)	H-2a (6) H-9 (5) H-9' (5)	H-9 (8)	H-5 (8)		H-9α (8)	H-5 (2) H-1 (2)	H-9α (7)	H-2 (8) H-9α (8)	H-2 (10) H-9\$ (10)
H-15	H-6 (10) H-8 (6)	H-6 (12) H-8 (7)	H-8 (19)	H-5 (8)		H-6 (6) H-8 (8)	H-6 (12)	H-2 <i>β</i> (12)	H-5 (15)		H-5 (10)	H-5 (10)	H-5 (12)	H-8 (12)	H-3 <i>β</i> (10) H-8 (12)	H-3 <i>f</i> ) (6)
ОН	·													H-5 (10) H-15	H-5 (8)	H-5 (5)

Table 8. NOE data for sesquiterpene lactones (intensities in parentheses)

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<sup>1</sup>H NMR spectra (Table 7). As in the case of 9a-9c the configuration of the allenic bond in 20 and 21 could be assigned as shown. The configuration of the  $\Delta^5$ -double bond in 21 previously not written clearly [20] also is E. The structures of the cyclobutane derivatives 22b and 23 were deduced by spin decoupling which allowed the assignment of all signals. The presence of an allylic coupling between H-5 and H-2 as well as of homoallylic couplings between H-5 and H-3 and H-8 and H-13 allowed combination of the obtained sequences. The stereochemistry was established by NOE difference spectroscopy. A clear NOE between H-15 and H-8 in the case of 22b required a syn-configuration. The corresponding anti-stereochemistry of 23 could be deduced from the NOEs between H-14, H-2 and H-9\beta and between H-15 and H-2β. In both cases a NOE between the hydroxyl proton and H-5 indicated a  $4\beta$ -methyl group.

Reaction of 2b with base gave first 17a and by hydrolysis 17b. The structures followed from the <sup>1</sup>H NMR spectra (Table 7). On standing in solution in daylight 17a was transformed to a mixture of the isomeric cyclobutane derivatives 18 and 19. All data of 18 were identical with those of a naturally occurring bourbonenolide (13 of ref. [10]). Additional NOE difference spectroscopy results showed that the proposed stereochemistry of this substance as well as its congeners numbered 12-14 in lit. [10] have to be revised (see Table 9). Thus clear NOEs were observed between H-5. H-8 and H-15 and between H-14 and H-9a (The configuration proposed for compounds numbered 18a and 18b in lit. [20] could be established by clear NOEs between H-14 and H-1 (7%), H-5 (2%) and H-9 $\beta$  (6%), between H-15 and H-5 (8%) as well as between H-1 and H-5 (10%)). The isomer 19 showed NOEs between H-5, H-14 and H-15. Furthermore a typical 6 Hz W-coupling between H-1 and H-5 indicated a cyclobutane derivative.

Compounds 18 and 19 obviously are formed by a photochemically induced [2+2]-cyclo addition. The <sup>1</sup>H NMR spectrum of 17a showed that this lactone was present in two conformations with different proportions

in chloroform and benzene. NOE difference spectroscopy allowed the assignment of the nature of the conformers. In the preferred one the C-10 methyl is below and the lactone ring above the plane while in the second conformer both are below the plane. As the relative amounts of the isomers obtained is influenced by the solvent most likely the two conformers are transformed to the different isomers. This assumption nicely agree with inspection of models.

The observed base catalysed transformations of several sesquiterpene lactones from *Vernonia* species also may be of interest in connection with their biogenesis. Most likely nearly all types of sesquiterpene lactones reported from *Vernonia* species are biogenetically closely related. Accordingly, it is very likely that the stereochemistry of the lactones is always the same at C-4  $(4\beta$ -methyl), C-8

Scheme 3

Table 9. Summary of sesquiterpene lactone formulae which have had to be amended

New overall structure (this paper)	Formulae numbers in previous papers (lit. references in brackets)*						
3	1-4 (1α-OR) [2], 5/6 [2]						
5	4 [5]						
Sa	36† [4], 7/8 [5], 14† [12], 7† [14]						
86	10/11 [2], 21-23 [4], 29-31 [4]						
84	8-13 [9], 13e-13c [12], 3/4 [14], 3-7 [15]						
	1/2 [16], 1g/1h [17], 1a-1f [18]						
18	12-14 [10]						
20	<b>4</b> ‡ [19], <b>15</b> ‡ <b>[2</b> 0], <b>16</b> [20]						
21	13 [5], 17 [20]						

<sup>\*</sup>Other authors have also pointed out in print that these structures were in error, see refs [17, 26].

 $(8\alpha\text{-OR})$ , C-10  $(10\alpha\text{-methyl},^{\bullet})$  but  $10\beta\text{-methyl}$  with oxygen function at C-10), C-6  $(6\beta\text{-H})$  if present) and E-configuration of the  $\Delta^5$ -double bond.

The isolation of 6a from Vernonia arctioides may be of interest as this species is placed in the section Leiboldia [23] which together with Lepidonia are proposed to be ancient relictual elements within the New World Vernonias. From V. jonesii, however, which belongs to the same group of more primitive species, a bis-epoxide, which lacks the 7(11)-double bond, was isolated [24]. A detailed discussion of biogenetic relationships of the sesquiterpene lactones in the tribe Vernonieae will be published elsewhere.

## **EXPERIMENTAL**

The air dried plant material was extracted with MeOH-Et<sub>2</sub>O-petrol (1:1:1) and the extracts obtained after treatment with MeOH to remove long chain hydrocarbons were separated in the usual way [25] first by CC (SiO<sub>2</sub>) followed by TLC (SiO<sub>2</sub>, PF 254) and in part by HPLC (RP 8, flow rate, 3 ml/min, ca 100 bar).

The extract of the aerial parts of Vernonia patens (200 g), collected in Peru, voucher RMK 9148, gave a polar CC fraction (Et<sub>2</sub>O, Et<sub>2</sub>O-MeOH, 9:1), which was separated again by CC affording three fractions: 1 (Et<sub>2</sub>O-petrol, 1:1), 2 (Et<sub>2</sub>O) and 3 (Et<sub>2</sub>O-MeOH, 19:1). TLC of 1 (Et<sub>2</sub>O-petrol, 3:1) gave 150 mg 4e ( $R_f$  0.61) and a mixture, which by repeated TLC (CHCl<sub>3</sub>, five developments) afforded 40 mg 3 ( $R_f$  0.45), 20 mg 4e ( $R_f$  0.48) and 6 mg 4d ( $R_f$  0.38). TLC of fraction 2 (Et<sub>2</sub>O) gave 30 mg 8b ( $R_f$  0.66) and a mixture, which by repeated TLC (CHCl<sub>3</sub>-Et<sub>2</sub>O, 3:1) gave 10 mg 5b ( $R_f$  0.33) and 2 mg 5a ( $R_f$  0.28). TLC of fraction 3 (CHCl<sub>3</sub>-Et<sub>2</sub>O, 3:1, three developments) afforded 5 mg 8a ( $R_f$  0.45), 3 mg 10c ( $R_f$  0.4), 2 mg 10c ( $R_f$  0.38) 2 mg 10a ( $R_f$  0.35) and 3 mg 7a ( $R_f$  0.21).

The extract of the aerial parts of Vernonia jalcana (500 g. collected in Peru, voucher RMK 9250) gave a crude polar CC fraction (Et<sub>2</sub>O and Et<sub>2</sub>O-MeOH, 9:1) which after repeated CC afforded three fractions: 1 (Et<sub>2</sub>O), 2 (Et<sub>2</sub>O-MeOH, 9:1) and 3 (Et<sub>2</sub>O-MeOH, 3:1). HPLC (MeOH-H<sub>2</sub>O, 3:2) of fraction 1 gave 10 mg 8d (R<sub>1</sub> 4.3 min), 4 mg 8f (R<sub>2</sub> 6.0 min), 5 mg 8e (R<sub>3</sub> 7.8 min), 3.5 mg 8g (R<sub>4</sub> 11.2 min) and 2 mg apigenin 7-methyl

ether. HPLC (MeOH- $H_2O$ , 1:1) of fraction 2 gave 2 mg 8c ( $R_1$  4.5 min), 10 mg 12b ( $R_1$  5.9 min), 52 mg 4a ( $R_1$  6.2 min), 51 mg 8d ( $R_2$  6.9 min), 21 mg 8h ( $R_2$  7.8 min), 30 mg 6a ( $R_1$  10.5 min), 6 mg 8f ( $R_2$  10.9 min), 15 mg 4d ( $R_2$  14.0 min), 3 mg 2a ( $R_2$  14.9 min), 5 mg 4b ( $R_2$  16.2 min), 4.7 mg 8e ( $R_2$  17.7 min), 1.5 mg 11 ( $R_2$  20.4 min), 13 mg 6b ( $R_2$  27.0 min) and a mixture ( $R_2$  4.9 min), which by TLC (CH<sub>2</sub>Cl<sub>2</sub>- $C_4$ H<sub>6</sub>-Et<sub>2</sub>O, 2:2:1, five developments) gave 1.5 mg 10d ( $R_2$  0.85) and 3 mg 10b ( $R_2$  0.60). HPLC (MeOH- $H_2$ O, 3:2) of 40 mg of fraction 3 gave 2 mg 12a ( $R_2$  3.2 min) 6 mg 12b ( $R_2$  4.0 min) and an unresolved mixture. The remaining part of fraction 3 was acetylated (AcCl,  $C_6$ H<sub>3</sub>NMe<sub>2</sub>, 72 hr, 20°). HPLC (MeOH- $H_2$ O, 1:1) gave 8 mg 12b-acetate ( $R_2$  8.0 min), 3 mg 12a ( $R_2$  5.0 min), 15 mg 12a-acetate ( $R_2$  8.5 min), 4 mg 12c ( $R_2$  21.0 min) and 3 mg 12d-acetate ( $R_2$  22.0 min).

The aerial parts of *Vernonia cotoneaster* (130 g, collected in the Province Bahia, Brazil, voucher RMK 8783) gave by CC a polar fraction (Et<sub>2</sub>O, Et<sub>2</sub>O-MeOH, 9:1), which by TLC (Et<sub>2</sub>O-petrol, 3:1) afforded 15 mg 9c ( $R_f$  0.4), 3 mg 9a ( $R_f$  0.2), 12 mg 9b ( $R_f$  0.27) and 0.5 mg 1 ( $R_f$  0.3).

The extract of the aerial parts of Pseudoelephantopus spicatus (170 g, collected in Jamaica, voucher Jam 9) afforded a polar CC fraction (Et<sub>2</sub>O and Et<sub>2</sub>O-MeOH, 9:1) which gave after TLC (Et<sub>2</sub>O) a mixture and 180 mg 6a. HPLC of the mixture (MeOH-H<sub>2</sub>O, 7:3) gave 2 mg 13 (R, 2.0 min), 10 mg 6a (R, 2.2 min) and 2 mg 2a (R, 3.7 min).

Similarly the extract of the aerial part of Vernonia acuminata (170 g, collected in Jamaica, voucher Jam 19) gave 100 mg 6a, that of the aerial parts of V. arctioides (230 g, collected in Mexico, voucher TEX 15527) gave 20 mg 6a, that of the aerial parts of V. aurea (450 g, collected in the province Bahia, Brazil, voucher RMK 8793) gave 25 mg 6a, 100 mg 8i and 40 mg 8k, that of the aerial parts of V. brachiata (500 g, collected in Costa Rica) 15 mg 6a, that of the roots of V. mariana (100 g, collected in the province Bahia, Brazil, voucher RMK 8558) 3 mg 4d, 1 mg 5a and 1 mg 11\(\beta\),13-dihydroxy-11,13-dihydroeremanthin and that of the aerial parts of V. greggii (120 g, collected near Monterrey, Mexico, voucher 1643/84) 10 mg 6b and 10 mg 8l.

 $2\beta$ -Acetoxy -8 $\alpha$ , 13 -dihydroxygermacra - 1(10), 4,7(11) - trien-6 $\alpha$ , 12-olide (1). Colourless oil; CIMS m/z (rel. int.): 323 [M + 1]\* (100), 305 [323 - H<sub>2</sub>O]\* (21), 263 [323 - HOAc]\* (28), 245 [263 - H<sub>2</sub>O]\* (49).

8-Desacylmarginatin 8-O-acetate (2a). Colourless crystals, mp 112-114°; IR  $\nu_{\rm CHCl_2}^{\rm CHCl_2}$  cm $^{-1}$ : 1770 (y-lactone), 1750, 1230 (OAc); MS m/z (rel. int.): 364.152 [M]\* (1) (calc. for  $C_{19}H_{24}O_7$ : 364.152), 304 [M - HOAc]\* (3), 244 [304 - HOAc]\* (20), 216

 $<sup>\</sup>dagger \Delta^1$ .

<sup>\$4,5-</sup>Desoxy.

<sup>\*</sup>So far the only exceptions are compound 1 in ref. [3] and 1-6 in ref. [13].

[244 - CO]\* (60), 68 (100).

$$[\alpha]_{24}^{\perp} = \frac{589}{-35} \frac{578}{-35} \frac{546}{-41} \frac{436 \text{ nm}}{-72} \text{ (CHCl}_3; c 0.3).$$

Stilpnotomentolide 8-O-angelate (3). Colourless oil;  $IR v_{COL}^{COL} cm^{-1}$ : 1775 (γ-lactone), 1745, 1230 (OAc), 1725 (OCOR), 1710 (C=O); MS m/z (rel. int.); 420.178 [M]\* (3) (calc. for  $C_{22}H_{29}O_8$ : 420.178), 360 [M - HOAc]\* (5), 321 [M - OCOR]\* (3), 261 [321 - HOAc]\* (4), 260 [360 - RCOOH]\* (5), 83 [C<sub>4</sub>H<sub>7</sub>CO]\* (100), 55 [83 - CO]\* (95).

$$[\alpha]_{24}^{\perp} = \frac{589}{-77} \frac{578}{-80} \frac{546}{-92} \frac{436 \text{ nm}}{-161} \text{ (CHCl}_3; c 0.76).$$

8-Desacylvernonataloide 8-O-acetate (4a). Colourless crystals, mp 169-171°; IR  $v_{\text{COl}_3}^{\text{COl}_3}$  cm $^{-1}$ : 1780 (y-lactone), 1760, 1240 (OAc); MS m/z (rel. int.): 320.126 [M-AcOH]\* (4) (calc. for  $C_{17}H_{20}O_6$ : 320.126), 278 [320-ketene]\* (14), 260 [320-AcOH]\* (10), 69 (90), 57 (100).

$$[\alpha]_{24}^{\perp} = \frac{589}{+106} \frac{578}{+110} \frac{546}{+124} \frac{436 \text{ nm}}{+207} \text{ (CHCl}_3; c 1.25).$$

8-Desacylvernonataloide 8-O-isobutyrate (4b). Colourless oil; IR  $v_{\rm CHCl_2}^{\rm CHCl_3}$  cm  $^{-1}$ : 1770 (y-lactone), 1740, 1240 (OCOR, OAc); MS m/z (rel. int.); 348.157 [M - AcOH] \* (1.5) (calc. for C<sub>19</sub>H<sub>24</sub>O<sub>6</sub>: 348.157), 320 [M - RCO<sub>2</sub>H] \* (1), 278 [348 - Me<sub>2</sub>C=C=O] \* (3), 260 [348 - RCO<sub>2</sub>H] \* (6), 71 [C<sub>3</sub>H<sup>2</sup>CO] \* (100).

8-Desacylvernonataloide 8-O-angelate (4c). Colourless crystals, mp 126.5°; IR  $v_{\text{End}}^{\text{CC}}$  cm<sup>-1</sup>: 1775 (y-lactone), 1750, 1230 (OAc), 1725 (OCOR); MS m/z (rel. int.): 420.178 [M]\* (1) (calc. for  $C_{22}H_{20}O_0$ : 420.178), 360 [M - HOAc]\* (2), 321 [M - OCOR]\* (2), 320 [M - RCOOH]\* (17), 261 [321 - HOAc]\* (3), 260 [320 - HOAc]\* (4), 83 [C<sub>6</sub>H<sub>7</sub>CO]\* (90), 55 [83 - CO]\* (100).

8a-(2-Methylacryloyloxy)-compactifloride (5a). Colourless oil; IR  $v_{\rm CCL}^{\rm CCL}$  cm<sup>-1</sup>: 3600 (OH), 1770 (y-lactone), 1750 (OAc), 1720 (C=CCO<sub>2</sub>R); CIMS m/z (rel. int.); 407 [M+1]\* (4), 389 [407 - H<sub>2</sub>O]\* (3), 321 [407 - RCO<sub>2</sub>H]\* (7), 261 [321 - HOAc]\* (21), 243 [261 - H<sub>2</sub>O]\* (21).

8 $\alpha$ -Angeloyloxycompactifloride (5b). Colourless oil; IR  $\nu_{\text{CML}}^{\text{CCL}}$  cm<sup>-1</sup>: 1770 ( $\gamma$ -lactone), 1750, 1235 (OAc), 1725 (OCOR); MS m/z (rel. int.): 360.157 [M - HOAc]\* (2) (calc. for  $C_{10}H_{14}O_4$ : 360.157), 260 [360 - RCOOH]\* (2), 83 [ $C_4H_7\text{CO}$ ]\* (90), 55 [83 - CO]\* (100).

$$[\alpha]_{24}^{\perp} = \frac{589}{-99} \frac{578}{-105} \frac{546}{-120} \frac{436 \text{ nm}}{-232} \text{ (CHCl}_3; c 0.58).$$

$$[\alpha]_{24}^{1} = \frac{589}{+37} \frac{578}{+40} \frac{546}{+46} + \frac{436 \text{ nm}}{101} \cdot \text{(CHCl}_{3}; c 0.2).$$

15 mg 3 were stirred for 2 hr at 60° in 0.5 ml dioxane and 0.1 ml 1 N  $\rm K_2CO_3$ . Usual work-up and TLC (Et<sub>2</sub>O) gave 10 mg 7a, identical with the natural product, and 3 mg 7b. IR  $\nu_{\rm max}^{\rm CCl}$  cm<sup>-1</sup>: 3600 (OH), 1770 (y-lactone), 1725 (OCOR), 1710 (C=O); MS m/z (rel. int.): 378 [M]\* (0.5), 360 [M - H<sub>2</sub>O]\* (0.5), 278 [M - RCOOH]\* (2) (calc. for  $\rm C_{15}H_{10}O_3$ : 278.115), 260 [278 - H<sub>2</sub>O]\* (4), 83 [C<sub>4</sub>H<sub>7</sub>CO]\* (100), 55 [83 - CO]\* (86).

8a-Angeloyloxy-10a-hydroxy-1-desoxyhirsutinolide 13-Oacetate (8a). IR v<sup>CCL</sup><sub>max</sub> cm<sup>-1</sup>: 3600 (OH), 1770 (y-lactone), 1745, 1230 (OAc), 1720 (OCOR); MS m/z (rel. int.); 420.178 [M]\* (3) (calc. for  $C_{12}H_{24}O_9$ : 420.178), 360 [M - HOAc]\* (8), 260 [360 - RCOOH]\* (40), 218 [260 -  $C_2H_2O$ ]\* (75), 83 [ $C_4H_7CO$ ]\* (100), 55 [83 - CO]\* (100).

$$[\alpha]_{24}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \,\mathrm{nm}}{+24 \ +26 \ +33 \ +83}$$
 (CHCl<sub>3</sub>; c 0.33).

8 $\alpha$ -Angeloyloxyhirsutinolide 13-O-acetate (8b). Colourless crystals, mp 125°; IR  $\nu_{\rm ECL}^{\rm CCL}$  cm $^{-1}$ : 1770 ( $\gamma$ -lactone), 1750, 1230 (OAc), 1715 (OCOR); MS m/z (rel. int.); 420.178 [M]\* (3) (calc. for  $C_{22}H_{28}O_8$ : 420.178), 360 [M - HOAc]\* (1), 260 [360 - RCOOH]\* (8), 83 [ $C_4H_7CO$ ]\* (100), 55 [83 - CO]\* (76).

$$[\alpha]_{24}^{\perp} = \frac{589 \quad 578 \quad 546 \quad 436 \,\mathrm{nm}}{+88 \ +95 \ +112 \ +232}$$
 (CHCl<sub>3</sub>; c 0.4).

8 $\alpha$ -Acetoxy-10 $\alpha$ -hydroxyhirsutinolide (8c). Colourless oil; IR  $\nu$ CHCl<sub>2</sub> cm<sup>-1</sup>: 3521 (OH), 1765 (y-lactone), 1750, 1230 (OAc); MS m/z (rel. int.): 354.1315 [M]\* (6) (calc. for  $C_{17}H_{22}O_{6}$ : 354.1315), 336 [M  $-H_{2}O$ ]\* (1), 318 [336  $-H_{2}O$ ]\* (1), 59 (100). 8 $\alpha$ -Acetoxy-10 $\alpha$ -hydroxy-13-O-methyl hirsutinolide (8d). Colourless oil; IR  $\nu$ CCl<sub>2</sub> cm<sup>-1</sup>: 3540 (OH), 1770 (y-lactone), 1750 (OAc); MS m/z (rel. int.): 368.147 [M]\* (8) (calc. for  $C_{16}H_{24}O_{6}$ : 368.147), 353 [M - Me]\* (6), 336 [M - MeOH]\* (2), 309 [M - OAc]\* (4), 308 [M - AcOH]\* (3.5), 276 [308 - MeOH]\* (16), 55 (100).

8 $\alpha$ -Methacryloyloxy-10 $\alpha$ -hydroxy-13-O-methyl hlrsutinolide (8e). Colourless oil; IR  $\nu_{COL}^{COL}$  cm  $^{-1}$ : 3540 (OH), 1770 (y-lactone), 1730 (OCOR); MS m/z (rel. int.); 394.163 [M]\* (0.2) (calc. for  $C_{20}H_{26}O_6$ : 394.163), 377 [M - OH]\* (0.3), 309 [M - OCOR]\* (0.8), 308 [M - RCO<sub>2</sub>H]\* (0.6), 276 [308 - MeOH]\* (5), 234 (20), 69 [ $C_3H_3$ CO]\* (100).

8a-Acetoxy-10a-hydroxy-1,13-bis-O-methyl hirsutinolide (Bf). Colourless oil; IR  $v_{max}^{CCL}$  cm<sup>-1</sup>: 3540 (OH), 1770 (y-lactone), 1750, 1230 (OAc); MS m/z (rel. int.): 382.163 [M]\* (2.5) (calc. for  $C_{19}H_{26}O_{g}$ : 382.163), 351 [M - OMe]\* (6), 350 [M - MeOH]\* (2.5), 290 [350 - AcOH]\* (10), 248 [290 -  $C_{2}H_{2}O$ ]\* (40), 232 (85), 55 (100).

8a-Methacryloyloxy-10a-hydroxy-1,13-bis-O-methyl hirsutinolide (8g). Colourless oil; IR  $v_{---}^{CCL}$  cm<sup>-1</sup>: 3580 (OH), 1780 (y-lactone), 1740 (OCOR); MS m/z (rel. int.); 408.178 [M]\* (1) (calc. for  $C_{21}H_{28}O_8$ : 408.178), 376 [M - MeOH]\* (0.5), 345 [377 - MeOH]\* (0.2), 290 [376 - RCO<sub>2</sub>H]\* (6), 248 (20), 232 (40), 69 [C<sub>3</sub>H<sub>5</sub>CO]\* (100).

Bis-desacetyl vernoallenolide (9a). Colourless oll;  $1R \ v_{max}^{CCL} \ cm^{-1}$ : 3600, 3400 (OH), 1945 (allene), 1760 (y-lactone); MS m/z (rel. int.): 262.121 [M]\* (0.5) (calc. for  $C_{13}H_{18}O_4$ : 262.121), 244 [M  $-H_2O$ ]\* (9), 211 [226 -Me]\* (6), 105 (100). 13-Desacetyl vernoallenolide (9b). Colourless oil;  $1R \ v_{max}^{CCL} \ cm^{-1}$ : 3450 (OH), 1750 (y-lactone, OAc); MS m/z (rel. int.): 244.110 [M -HOAc]\* (41) (calc. for  $C_{13}H_{18}O_3$ : 244.110), 226 [244  $-H_2O$ ]\* (24), 211 [226 -Me]\* (25), 183 [211 -CO]\* (21), 105 (100).

8\alpha-Angeloyloxy-4\alpha,5\alpha-epoxyjalcaguaianolide (1\bar{\text{0}}\alpha). Colourless oil which was purified by acetylation (AcCl, C<sub>6</sub>H<sub>3</sub>NMe<sub>2</sub>, 24 hr, 20°). TLC (Et<sub>2</sub>O) gave 1\bar{\text{0}}\alpha-diacetate; colourless oil; IR  $v_{\text{MAX}}^{\text{CCL}}$  cm  $^{-1}$ : 1780 (7-lactone), 1750 (OAc), 1730 (OCOR); MS m/z (rel. int.); 462.189 [M]\* (6) (calc. for C<sub>24</sub>H<sub>30</sub>O<sub>9</sub>: 462.189), 403 [M - OAc]\* (1), 402 [M - HOAc]\* (0.5), 360 [402 - ketene]\* (1), 342 [402 - HOAc]\* (0.5), 260 [360 - RCOOH]\* (3), 242 [342 - RCOOH]\* (4), 83 [C<sub>4</sub>H<sub>3</sub>CO]\* (100), 55 [83 - CO]\* (56).

8 $\alpha$ -Acetoxy-4 $\alpha$ ,5 $\alpha$ -epoxy-13-O-methyl jalcaguaianolide (100). Colourless oil;  $1R \ v_{\max}^{CC_1} \ cm^{-1}$ : 3400 (OH), 1770 (y-lactone), 1750, 1220 (OAc); MS (CI, isobutane) m/z (rel. int.): 353 [M + 1]\* (20) (calc. for  $C_1$ 7 $H_{24}O_7$  + 1), 335 [M + 1 -  $H_2O$ ]\* (100), 293 [M + 1 - AcOH]\* (25), 261 [293 - MeOH]\* (16), 243 [261 -  $H_2O$ ]\* (50).

8 $\alpha$ -Angeloyloxy-4 $\alpha$ ,5 $\alpha$ -epoxy-13-O-methyl jalcagualanolide (10c). Colourless oil; IR  $\nu$  CCl<sub>2</sub> cm<sup>-1</sup>: 3600 (OH), 1780 (y-lactone), 1735 (OCOR); MS m/z (rel. int.); 392 [M]\* (0.5), 360.155 [M-MeOH]\* (3) (calc. for  $C_{10}H_{24}O_4$ : 360.155), 260 [360-RCOOH]\* (4), 83 [ $C_4H_7$ CO]\* (100), 55 [83 - CO]\* (50). 8 $\alpha$ -Acetoxy-4 $\alpha$ ,5 $\alpha$ -epoxyjalcagualanolide 13-O-acetate (10d). Colourless oil; IR  $\nu$  CCl<sub>2</sub> cm<sup>-1</sup>: 3400 (OH), 1775 (y-lactone), 1755, 1230 (OAc); MS (Cl. isobutane) m/z (rel. int.); 381 ( $C_{19}H_{25}O_4$ ) [M+1]\* (10), 363 [M+1- $H_2O$ ]\* (20), 321 [363 - ketene]\* (40), 303 [363 - AcOH]\* (10), 243 [303 - AcOH]\* (100).

8\alpha-Angeloyloxy-4\alpha,5\alpha-epoxyjalcaguaianolide 13-O-acetate (10e). Colourless oil, which was purified as its acetate (10a-diacetate) which was identical with the acetylation product of 10a (see above).

 $\begin{array}{lll} & 4\alpha\text{-}Hydroxy\text{-}8\alpha\text{-}isobutyroyloxy\text{-}4.5\text{-}dihydro\text{-}5.6\text{-}dehydro\text{-}10,13\text{-}bis\text{-}O\text{-}methyljalcaguaianolide} & (11). & \text{Colourless} & \text{oil;} \\ & \text{IR $v_{\text{CHA}}^{\text{CHC}_3}$ cm$^{-1}$: 3500 (OH), 1770 (y-lactone), 1740 (OCOR);} \\ & \text{CIMS } m/z \text{ (rel. int.): 395 } [M+1]^* \text{ (25), 377 } [M+1-H_2O]^* \\ & \text{(20), 363 } [M+1-\text{MeOH}]^* \text{ (50), 275 } [363-\text{RCO}_2\text{H}]^* \text{ (50),} \\ & 243 } [275-\text{MeOH}]^* \text{ (55), 73 (100).} \end{array}$ 

Vernojalcanolide 8-O-acetate (12a). Colourless oil; IR  $v_{col}^{CCl_{+}}$  cm<sup>-1</sup>: 3520, 3400 (OH), 1770 (y-lactone); 1745, 1240 (OAc); MS m/z (rel. int.): 378.131 [M - HOAc,  $H_{2}O$ ]\* (8) (calc. for  $C_{19}H_{22}O_{6}$ : 378.131), 336 [M - ketene]\* (8), 318 [M - AcOH]\* (22), 276 [318 - ketene]\* (55), 258 [318 - AcOH]\* (30), 99 [ $C_{3}H_{1}O_{2}$ ]\* (90), 55 (100).

12n-acetate. Colourless oil; (<sup>1</sup>H NMR data see Table 5); IR  $v_{\text{col}}^{\text{CCl}_4}$  cm<sup>-1</sup>: 3560, 3321 (OH), 1770 (y-lactone), 1740-1760, 1210-1260 (OAc); MS m/z (rel. int.); 480.163 [M - AcOH]\* (3) (calc. for  $C_{23}H_{24}O_{11}$ : 480.163), 420 [480 - AcOH]\* (40), 378 [420 - ketene]\* (20), 318 [378 - AcOH]\* (40), 276 [318 - ketene]\* (100), 258 [318 - AcOH]\* (75), 99.045 [ $C_3H_7O_2$ ]\* (75) (calc. for  $C_3H_7O_2$ : 99.045); CIMS: 541 [M+1]\* (15) ( $C_{23}H_{33}O_{13}$ ), 481 [M+1-AcOH]\* (100), 421 [481 - AcOH]\* (75).

13-O-Methylvernojalcanolide 8-O-acetate (12b). Colourless oil; IR  $v_{\text{max}}^{\text{CCL}}$  cm  $^{-1}$ : 3580, 3460 (OH), 1785 (7-lactone), 1760, 1260 (OAc); MS m/z (rel. int.): 438.1526 [M - MeOH]\* (2) (calc. for  $C_{21}H_{26}O_{10}$ : 438.1526), 378 [438 - AcOH]\* (20), 318 [378 - AcOH]\* (40), 276 [318 - ketene]\* (95), 258 [318 - AcOH]\* (60), 99 (100);  $^{13}C$  NMR (CDCl<sub>3</sub>): (C-1-C-15): 89.2 s, 30.6 t, 35.9 t, 73.3 s, 76.1 d, 76.9 s, 157.1 s, 64.9 d, 34.1 t, 84.4 s, 129.7 s, 168.9 s, 63.1 t, 19.1 q, 23.4 q, OMe: 58.5 q, OAc: 22.9 q, 21.1 q, 20.4 q, 171.3 s, 171.1 s, 170.4 s.

12b-acetate. Colourless oil; (<sup>1</sup>H NMR see Table 5). <sup>13</sup>C NMR (CDCl<sub>3</sub>) (C-1 C-15): 88.9 s, 29.9 t, 32.5 t, 84.1 s, 73.1 d, 75.9 s, 156.8 s, 64.6 d, 33.9 t, 84.4 s, 130.6 s, 168.6 s, 63.0 t, 19.6 q, 18.7 q, OMe: 58.5 q, OAc: 22.4 q, 22.1 q, 21.1 q, 20.1 q, 170.7 s, 170.6 s, 169.4 s, 168.9 s.

13-O-Methylvernojalcanolide 8-O-isobutyrate (12c). Colourless oil;  $1R \times_{max}^{CHCl_3}$  cm<sup>-1</sup>: 3500 (OH), 1770 (y-lactone), 1755-1740, 1250-1210 (OCOR, OAc); MS m/z (rel. int.); 378.131 [M - MeOH, RCO<sub>2</sub>H]\* (5) (calc. for C<sub>1</sub>+H<sub>22</sub>O<sub>9</sub>: 378.131), 318 [378 - AcOH]\* (15), 276 [318 - ketene]\* (50), 258 [318 - AcOH]\* (40), 99 (60), 71 [C<sub>3</sub>H<sub>7</sub>CO]\* (100).

12d-acetate. Colourless oil; IR  $v_{max}^{CCl_{*}}$  cm<sup>-1</sup>: 3460 (OH), 1770 (y-lactone), 1750, 1260 (OCOR, OAc), CIMS m/z (rel. int.): 539 [M + 1] \* (C<sub>26</sub>H<sub>35</sub>O<sub>12</sub>) (8), 507 [M + 1 - MeOH] \* (6), 479 [M + 1 - AcOH] \* (25), 421 [507 - RCO<sub>2</sub>H] \* (100), 419 [479 - AcOH] \* (35).

Pseudoelephantopide 8-O-methocrylate (13). Colourless oil;  $IR v_{COL}^{COL}$  cm<sup>-1</sup>: 1780 (y-lactone), 1740 (OCOR), 1720 (C=O); MS m/z (rel. int.); 333 [M - CHO]\* (5), 276.100 [M - RCOOH]\* (3) (calc. for  $C_{13}H_{16}O_3$ ; 276.100), 69 [ $C_3H_3CO$ ]\* (100).

Reaction with K<sub>2</sub>CO<sub>3</sub>. (1) 15 mg 4c were stirred in 3 ml dioxane for 1 hr at 60° with 0.5 ml 0.1 N K<sub>2</sub>CO<sub>3</sub> soln. TLC (Et<sub>2</sub>O, three

developments) gave 3 mg 10e, identical with the natural compound, and 2.5 mg 14  $(R_f 0.31)$  (<sup>1</sup>H NMR see Table 7). Similarly 3 mg 50 gave 2 mg 10e  $(R_f 0.42)$ .

(2) 30 mg 6a in 3 ml dioxane were stirred for 2 hr at 60° with 0.5 ml 1 N K<sub>2</sub>CO<sub>3</sub> soln. TLC (Et<sub>2</sub>O) gave 20 mg 16b ( $R_f$  0.48), colourless crystals, mp 137°; IR  $\nu_{\max}^{CCL}$  cm<sup>-1</sup>: 3600 (OH), 1780 (y-lactone), 1750 (OAc); MS m/z (rel. int.); 336.121 [M]\* (2) (calc. for C<sub>17</sub>H<sub>20</sub>O<sub>7</sub>: 336.121), 318 (2) [M - H<sub>2</sub>O]\*, 276 (12) [M - HOAc]\*, 218 (70), 163 (100), 99 (90).

(3) Similarly 50 mg \$1 gave after stirring for 90 min by TLC (Et<sub>2</sub>O) 30 mg 16b and 10 mg starting material.

(4) 50 mg 20 were stirred in 5 ml dioxane 90 min at 60° with 1 ml 0.1 N  $K_2$ CO<sub>3</sub>. TLC (Et<sub>2</sub>O) gave 30 mg 21 ( $R_f$  0.52), 10 mg of a mixture ( $R_f$  0.34) and 1.5 mg 22a ( $R_f$  0.28). Repeated TLC (CHCl<sub>3</sub>-Et<sub>2</sub>O, 2:1) of the mixture gave 7 mg 22b ( $R_f$  0.30) and 2.5 mg 23 ( $R_f$  0.37). 22b: Colourless oil; MS m/z (rel. int.): 362 [M]\* (0.5), 302.115 [M - HOAc]\* (14) (calc. for  $C_{17}H_{18}O_{3}$ : 302.165), 259 [302 - COMe]\* (45), 243 [302 - OAc]\* (50), 242 [302 - HOAc]\* (50), 199 (100). 23: Colourless oil; MS m/z (rel. int.): 362.137 [M]\* (4) (calc. for  $C_{19}H_{22}O_{7}$ : 362.137), 302 [M - HOAc]\* (40), 259 [302 - COMe]\* (70), 243 [302 - OAc]\* (66), 242 [302 - HOAc]\* (62), 199 (100).

(5) 25 mg 2b in 2 ml dioxane were stirred for 15 min with 0.6 ml 0.1 N  $K_2CO_3$  soln at 70°. TLC (CHCl<sub>3</sub>-Et<sub>2</sub>O, 1:1) gave 15 mg 17a and 5 mg 17b (<sup>1</sup>H NMR see Table 7). Acetate 17a on standing in  $C_6H_6$  at room temp. in daylight gave after TLC (CHCl<sub>3</sub>-Et<sub>2</sub>O, 1:1) 1 mg 18 ( $R_7$  0.65), identical with natural material and 10 mg 19 ( $R_7$  0.61), colourless oil; MS m/z (rel. int.); 344.162 [M]\* (1) (calc. for  $C_{20}H_{24}O_3$ ; 344.162), 262 [M -O-C=C(Me)CH=CH<sub>2</sub>]\* (4), 244 [M  $-RCO_2H$ ]\* (6), 218 [244 -CO]\* (6), 186 (18), 83 [ $C_4H_7CO$ ]\* (100), 55 [83 -CO]\* (70).

X-Ray analysis of **3b**. Crystals were orthorhombic, space group  $P2_12_12_1$ , unit cell parameters a=20.978 Å, b=12.175 Å, c=12.75 Å ( $\alpha$ ,  $\beta$ ,  $\gamma=19^\circ$ ). Data were collected by aid of a four-circle diffractometer Syntex P2 using MoK $\alpha$  radiation. Details will be published elsewhere.

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## REFERENCES

- 1. Seaman, F. C. (1982) Bot. Rev. 48, 121.
- Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1982) Phytochemistry 21, 1045.
- Bohlmann, F., Wallmeyer, M. and Jakupovic, J. (1982) *Phytochemistry* 21, 1445.
- Bohlmann, F., Brindöpke, G. and Rastogi, R. C. (1978) Phytochemistry 17, 475.
- Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1982) Phytochemistry 21, 695.
- 6. Bohlmann, F. and Zdero, C. (1982) Phytochemistry 21, 2263.
- Zabel, V., Watson, W. H., Mabry, T. J. and Padolina, W. G. (1980) Acta Crystallogr. B36, 3024.
- Padolina, W. G., Yoshioka, H., Nakatani, N., Mabry, T. J., Monti, S. A., Davis, R. E., Cox, P. J., Sim, G. A., Watson, W. H. and Wu, I. B. (1974) Tetrahedron 30, 1161.
- Bohlmann, F., Mahanta, P. K. and Dutta, L. N. (1979) *Phytochemistry* 18, 289.
- 10. Bohlmann, F., Singh, P., Borthakur, N. and Jakupovic, J.

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- (1981) Phytochemistry 20, 2379.
- Rogers, D., Moss, G. P. and Neidle, S. (1972 J. Chem. Soc. Commun. 142.
- Bohlmann, F., Müller, L., Gupta, R. K., King, R. M. and Robinson, H. (1981) Phytochemistry 20, 2233.
- Bohlmann, F., Ates(Gören) N. and Jakupovic, J. (1983) *Phytochemistry* 22, 1159.
- Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1981) Phytochemistry 20, 518.
- Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1979) Phytochemistry 18, 987.
- Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1983) Phytochemistry 22, 2863.
- 17. Herz, W. and Kulanthaivel, P. (1983) Phytochemistry 22, 1286.
- 18. Cowall, P. L., Cassady, J. M., Chang, C. and Kozlowski, J. F.

- (1981) J. Org. Chem. 46, 1108.
- Bohlmann, F., Gupta, R. K., Jakupovic, J., King, R. M. and Robinson, H. (1980) Liebigs Ann. Chem. 1904.
- Bohlmann, F., Jakupovic, J., Gupta, R. K., King, R. M. and Robinson, H. (1981) Phytochemistry 20, 473.
- Bohlmann, F., Singh, P., Zdero, C., Ruhe, A., King, R. M. and Robinson, H. (1982) Phytochemistry 21, 1669.
- Mabry, T. J., Abdel-Baset, Z., Padolina, W. G. and Jones, S. B. (1975) Biochem. Syst. Ecol. 2, 185.
- 23. Turner, B. L. (1981) Brittonia 33, 401.
- Gershenzon, J., Pfeil, R. M., Liu, Y. L., Mabry, T. J. and Turner, B. L. (1984) Phytochemistry 23, 777.
- Bohlmann, F., Zdero, C., King, R. M. and Robinson, H. (1984) Phytochemistry 23, 1979.
- Herz, W., Govindan, S. V. and Blount, J. F. (1981) J. Org. Chem. 46, 761.